



DPP-1386

M. Sc. (Sem. II) (Reg., Eve. & S.F.) Examination

April / May - 2016

Inorganic Chemistry : Paper - I

Time : 3 Hours]

[Total Marks : 70

**Instructions :**

(1)

नीचे दशांशवैल निशानीवाणी विगतो उत्तरवडी पर अवश्य कभवी. Fillup strictly the details of signs on your answer book.	Seat No. :
Name of the Examination :	<input type="text"/>
<input type="checkbox"/> M. Sc. (SEM. 2) (REG., EVE. & S.F.)	<input type="text"/>
Name of the Subject :	<input type="text"/>
<input type="checkbox"/> INORGANIC CHEMISTRY - 1	<input type="text"/>
Subject Code No. : <input type="text" value="1"/> <input type="text" value="3"/> <input type="text" value="8"/> <input type="text" value="6"/>	<input type="text"/>
Section No. (1, 2,.....) : <input type="text" value="NIL"/>	<input type="text"/>
	Student's Signature

- (2) Attempt all the questions.  
(3) Figures to the right indicate full marks.  
(4) Answer of all questions to be written in same answer sheet.

1 Answer briefly any **three** of the following : 18

- (a) What is diamagnetism ? Explain the origin of diamagnetism.  
(b) Describe the Gouy method for determining magnetic susceptibility of solid substances.  
(c) Give an account of temperature independent paramagnetism with suitable example.  
(d) Write a short note on diamagnetism and Pascal constant.

2 Answer briefly any **three** of the following : 18

- (a) What are metallic carbonyls ? Give classification of carbonyls.  
(b) Discuss the structure and bonding of  $\text{Os}_3(\text{CO})_{12}$   
(c) How is vibrational spectroscopy helpful in determination of geometries of metal nitrosyl ? Give your answer with suitable example.  
(d) Explain the preparation and properties of nitrosyl halides of Fe, Co and Ni.

**3** Answer briefly any **three** of the following : **18**

- (a) Give the classification and the limitations of inorganic polymers.
- (b) What is crystallinity ? Describe X-ray diffraction method for determining of percent crystallinity of polymer.
- (c) Explain molecular weight, number average and weight average in inorganic polymers.
- (d) Explain chemical and stereochemical variability in inorganic Polymers.

**4** Answer briefly any **three** of the following : **16**

- (a) Explain the terms :
    - (i) Pole strength
    - (ii) Magnetic susceptibility
    - (iii) Magnetic induction .
  - (b) What is EAN rule? Calculate EAN for any two from the following :
    - (i)  $\text{Fe}_2(\text{CO})_9$
    - (ii)  $\text{Ni}(\text{CO})_4$
    - (iii)  $\left[ \text{Co}^-(\text{CO})_3(\text{NO}^+) \right]^0$
  - (c) Write a short note on spin orbit coupling.
  - (d) Discuss, in brief, the structure, properties and uses of poly-phosphazenes.
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