



JB-3160

B. Sc. (Sem. IV) Examination

April/May – 2013

Applied Physics : Paper - V

(Crystallography & Material Science) (New Course)

Time : Hours]

[Total Marks : 50

Instructions :

(1)

नीचे दशांशों के निशानों के बिना उत्तरवही पर अवश्य लिखें। Fillup strictly the details of signs on your answer book.	Seat No. :
Name of the Examination :	<input type="text"/>
B. SC. (SEM. 4)	<input type="text"/>
Name of the Subject :	<input type="text"/>
APPLIED PHYSICS : PAPER - 5 (NEW)	<input type="text"/>
Subject Code No. : <input type="text"/> 3 <input type="text"/> 1 <input type="text"/> 6 <input type="text"/> 0	Student's Signature
Section No. (1, 2,.....): <input type="text"/> Nil	

- (2) Question one is compulsory.
- (3) Figures to the right indicate the total marks carried by the question.
- (4) Symbols used in the question paper have their usual meanings.
- (5) Students can use non-programmable scientific calculator.

- 1 Answer the following in brief : (any eight) 8
 - (i) What is meant by crystallography ?
 - (ii) Define the terms Unit cell, Primitive cell and Lattice parameters.
 - (iii) What are crystal planes ?
 - (iv) What are Bravais lattices ?
 - (v) What is drift velocity ?
 - (vi) What are magnetic materials ?
 - (vii) Write the uses of Hall effect.
 - (viii) Define the term electrical conductivity.
 - (ix) What is dielectric loss ?
 - (x) Write the name - types of SMAs.

- 2 (a) Explain a simple cubic unit cell and hence find the packing density of a simple cubic unit cell. 10

- (b) Calculate the interplanar spacing for (1 0 1) and (2 2 1) planes in a simple cubic lattice whose lattice constant is 0.42 nm. 4

OR

- 2 (a) Explain a FCC unit cell and hence find the packing density of a FCC unit cell. 10
(b) Sketch the following crystallographic planes for the cubic system $(0 \bar{1} 0)$, $(\bar{1} 0 0)$, $(0 0 \bar{1})$ and $(2 1 1)$. 4
- 3 (a) Describe with neat sketch the following unit cells. 10
(i) NaCl
(ii) Graphite.
(b) For a simple cubic lattice, lattice parameter is 2.04 \AA , calculate the spacing of the lattice plane (2 1 2). 4

OR

- 3 (a) What is meant by allotropy ? Describe the allotropy of carbon. 10
(b) The interplanar distance between the planes $(\bar{1} 1 1)$ in A1 (FCC structure) is 0.2388 nm. What is the lattice constant ? 4
- 4 Write short notes on : (any two) 14
(i) Frequency dependence of polarization
(ii) Application of SMAs
(iii) Ferromagnetic materials
(iv) Isotropy and Non-isotropy.